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for the reaction $\text{Br}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightleftharpoons 2\text{BrCl}(\text{g})$ explain the answer 2 b the graph shows that at $t = 120 \text{ s}$ the rates of both the forward and reverse

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up hydrated copper sulfate a colour change from blue to white and water vapour steam produced define the term dynamic equilibrium

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